



The Role of HRMS in Searching for Xenobiotic Compounds in Environmental Media

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US EPA

Office of Research and Development (ORD)

Center for Environmental Measurement and Modeling (CEMM)



August 3rd, 2020

NEMC The Environment in 2020: Past,
Present and Future

Compound identification basis



Foundation: First step
Critical for home
Necessary
Everything else is built upon it

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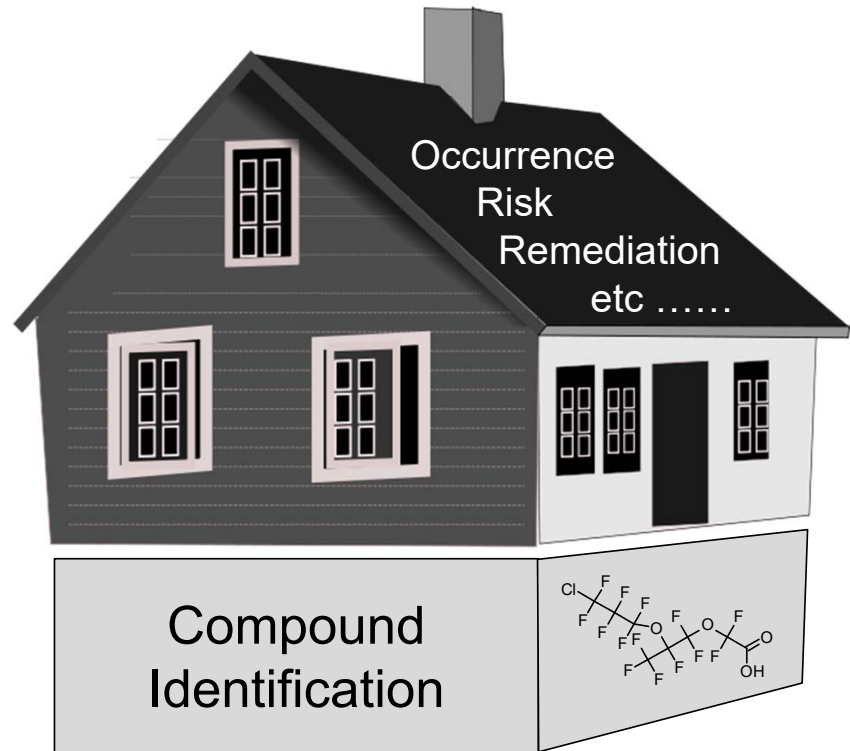
Compound
Identification



Compound identification basis



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Mass Spectrometry Resolution?

LR MS or MS/MS



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HRMS



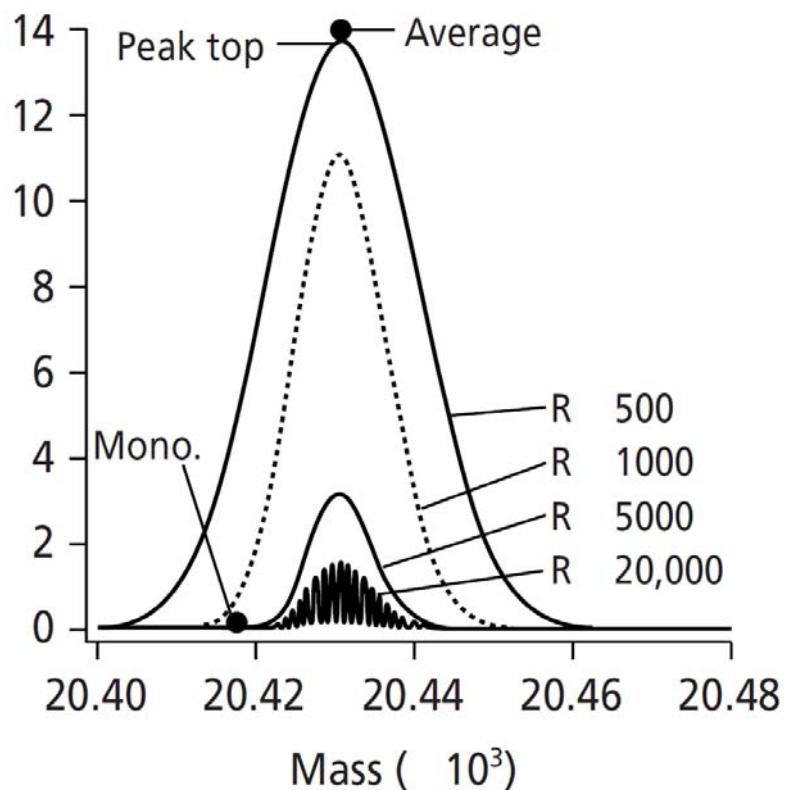
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Resolution is “the ability to “image” a detected ion in a mass spectrometer and differentiate it from any other”.

Balogh 2004 Debating Resolution and Mass

Accuracy LCGC NORTH AMERICA VOLUME 22 NUMBER 2

Mass Spectrometry Resolution?



Type	Resolving Power (FWHM)
FT-ICR-MS	1,000,000
FT-Orbitrap	100,000
High-Res-TOF	60,000
TOF	10,000
Quadrupole / IonTrap in UltraZoom mode	10,000
Quadrupole / Iontrap	1,000

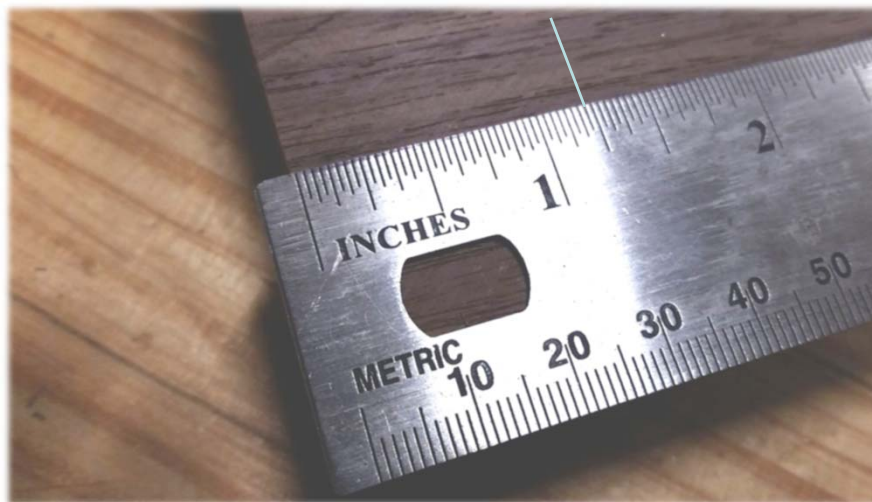
<https://fiehnlab.ucdavis.edu/projects/seven-golden-rules/mass-resolution>

Mass Accuracy

PFOA: MW 412.9 +/- 0.1 to 0.5 Da



MW 412.9964 +/- 0.0004 Da (1ppm)



photos by Mark Strynar

Mass accuracy is the ability to measure or calibrate the instrument response against a known entity. Usually expressed in parts per million (ppm), the measurement indicates the deviation of the instrument response from a known.

*Balogh 2004 Debating Resolution and Mass Accuracy
LCGC NORTH AMERICA VOLUME 22 NUMBER 2*

HRMS Measurable Parameters

Mass Accuracy

True mass	400.0000
Measured mass	400.0020
Difference	0.0020 or 2 mmu
Error	$\frac{0.002}{400} \times 10^6 = 5 \text{ ppm}$

Resolution (FWHM method)

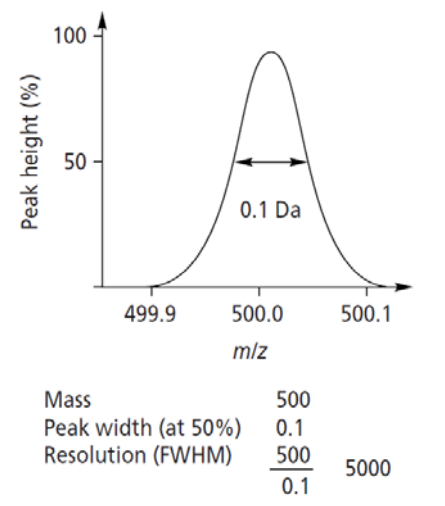


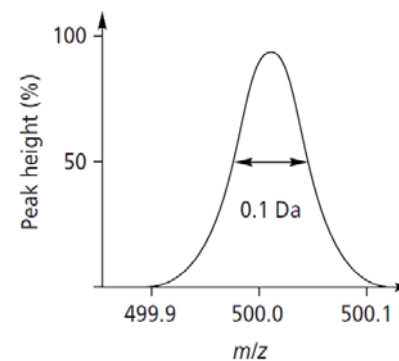
Figure 1: (Top) Mass accuracy determination and (bottom) the FWHM method for determining resolution for a mass spectrometer measured at a given ion.

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Resolution (FWHM method)



Mass	500
Peak width (at 50%)	0.1
Resolution (FWHM)	$\frac{500}{0.1}$ 5000

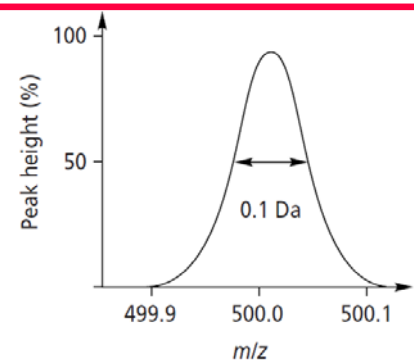
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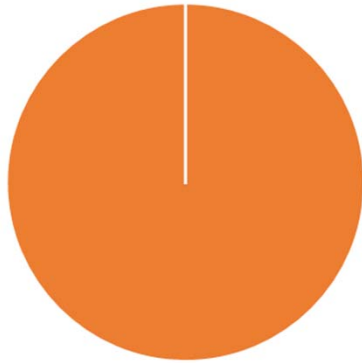
Resolution (FWHM method)



Mass	500
Peak width (at 50%)	0.1
Resolution (FWHM)	$\frac{500}{0.1} = 5000$

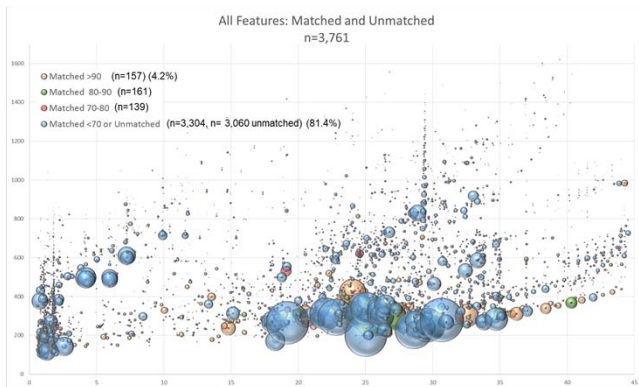
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What is our sample analysis comfort level?

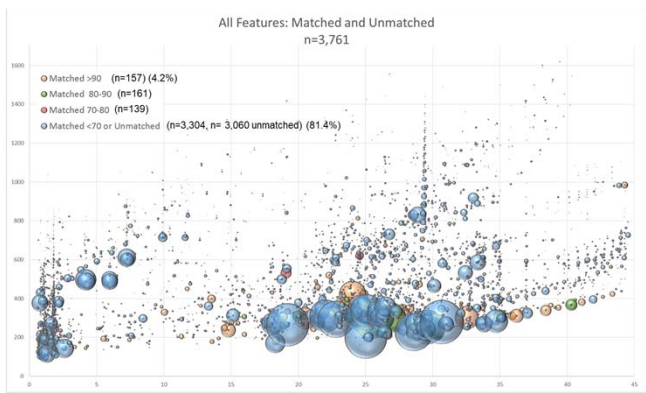
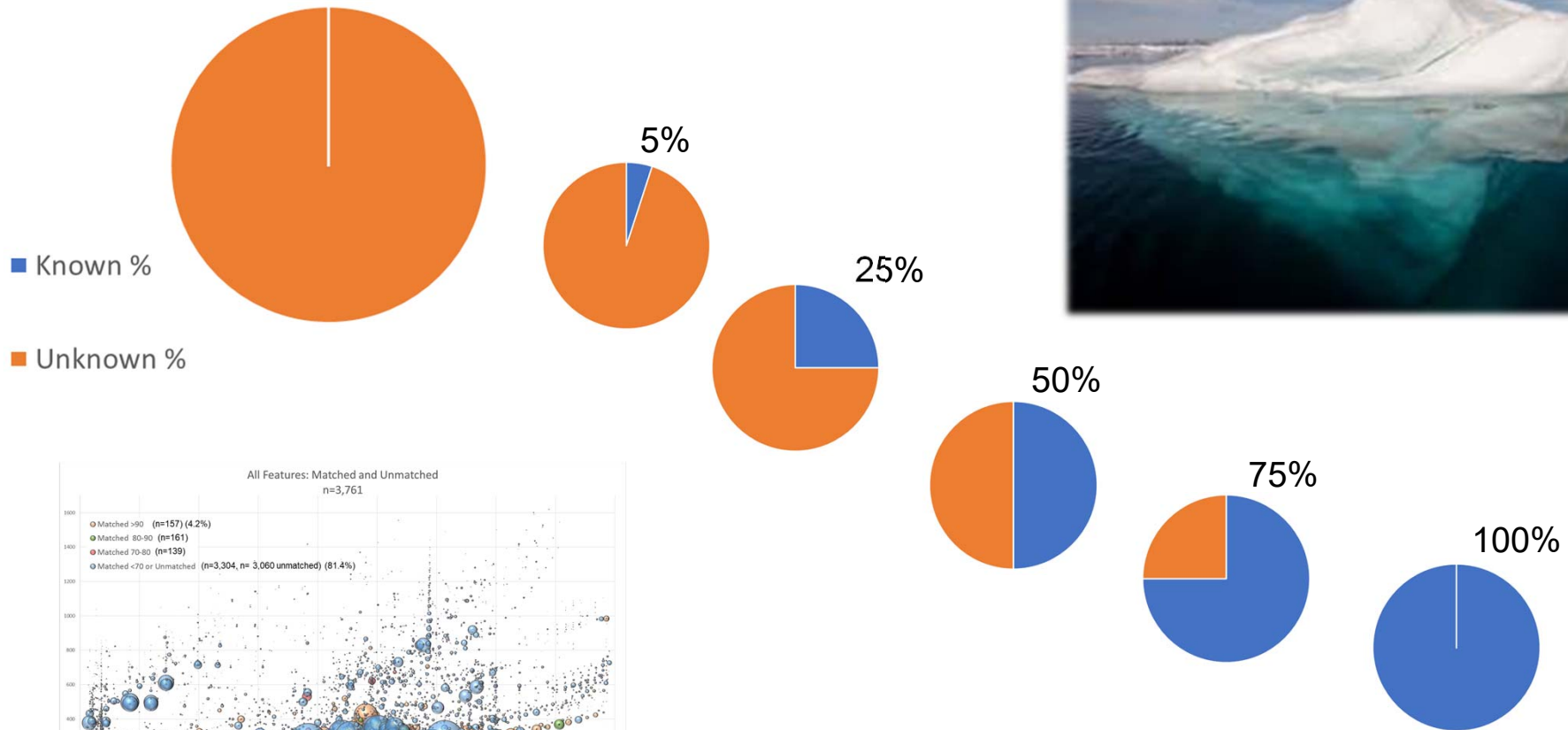


■ Known %

■ Unknown %

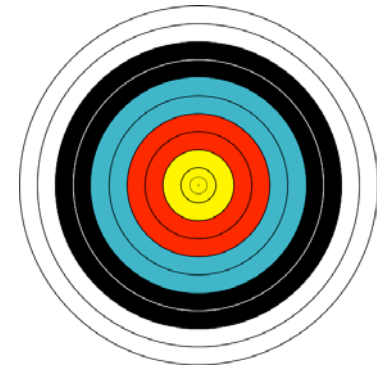


What is our sample analysis comfort level?



A few items defined

Targeted Analysis - analyzing for a suite of analytes with a validated method
e.g. How much Fipronil is in my water?



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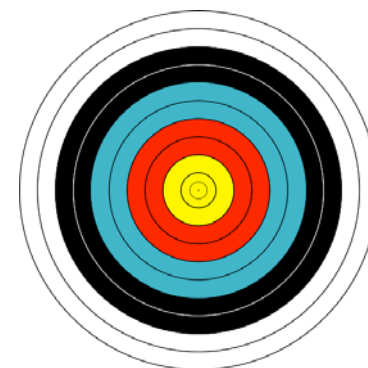
Non-Targeted Analysis (Untargeted Analysis) - no preconceived notion of chemical present

e.g. What chemicals are in my water?

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LRMS

HRMS



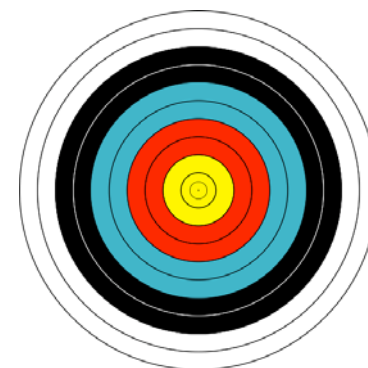
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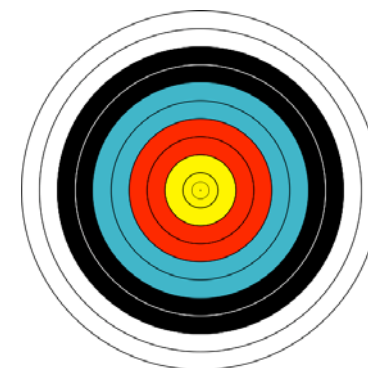
e.g. What chemicals are in my water?



Suspect Screening –
screening against a database
of chemicals

A few items defined

Targeted Analysis - analyzing for a suite of analytes with a validated method
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LRMS

HRMS



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Non-Targeted Analysis (Untargeted Analysis) - no preconceived notion of chemical present

e.g. What chemicals are in my water?



Suspect Screening –
screening against a database
of chemicals

Non-Targeted Analysis –
novel compound identification

Some key questions we are all asking:

How good is good enough?



Can we give quantitative values of these chemicals?

How will this data be used?

What are the important analytes?

What is my confidence in identification?

What if I miss something?

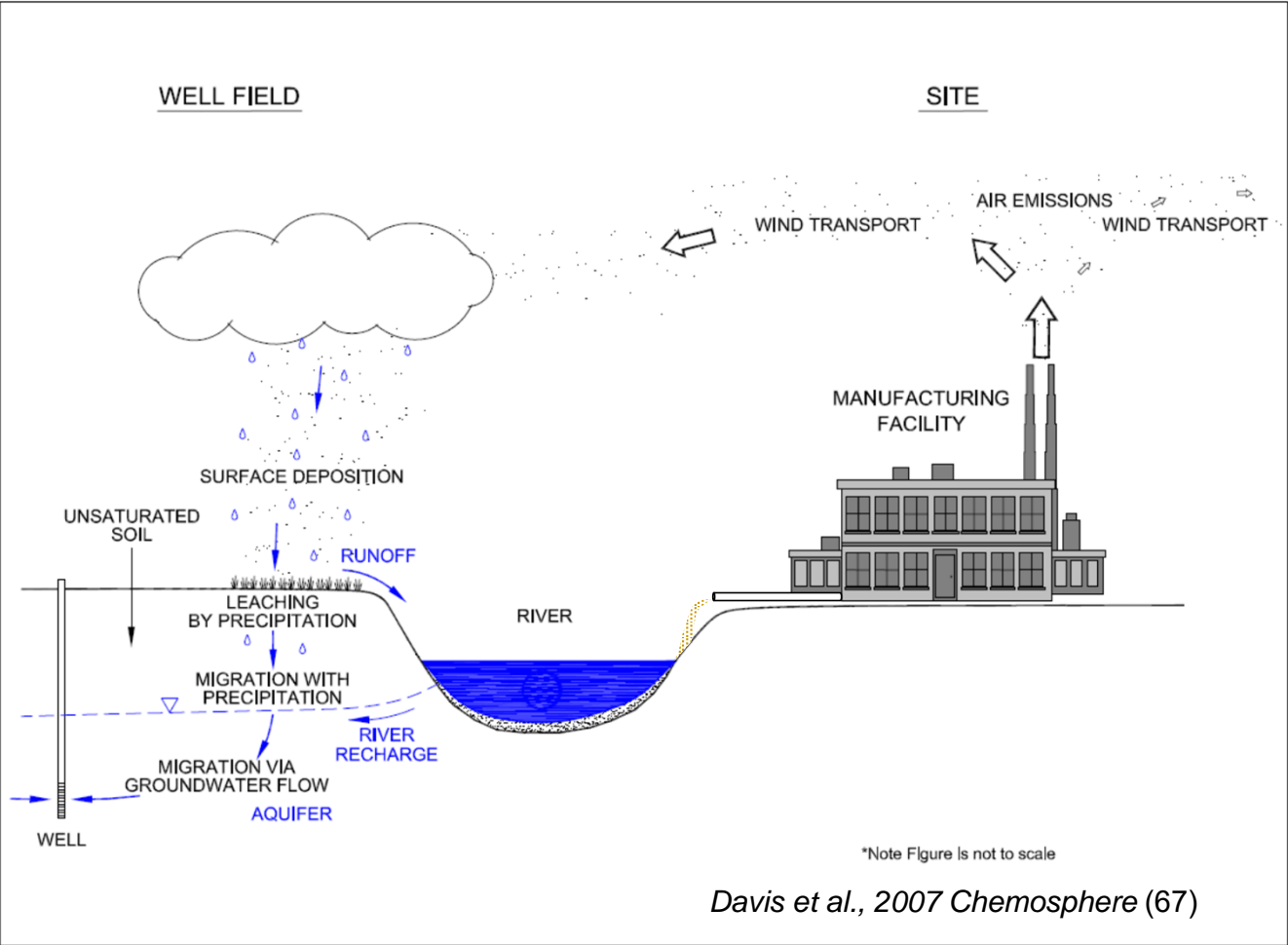
HRMS is A tool in the environmental/analytical chemists tool belt:

- a powerful tool**
- a very useful tool**
- not the only tool**

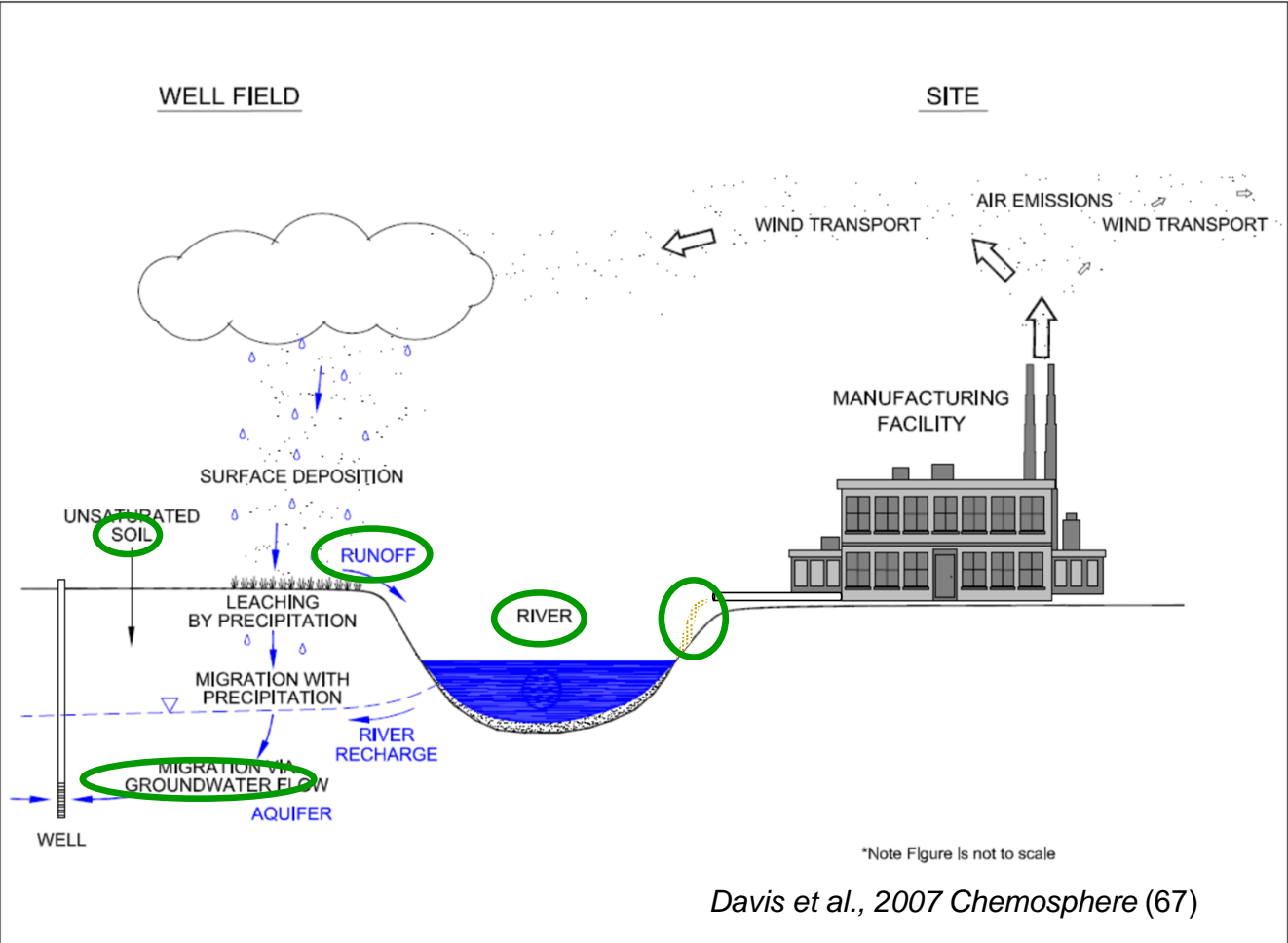


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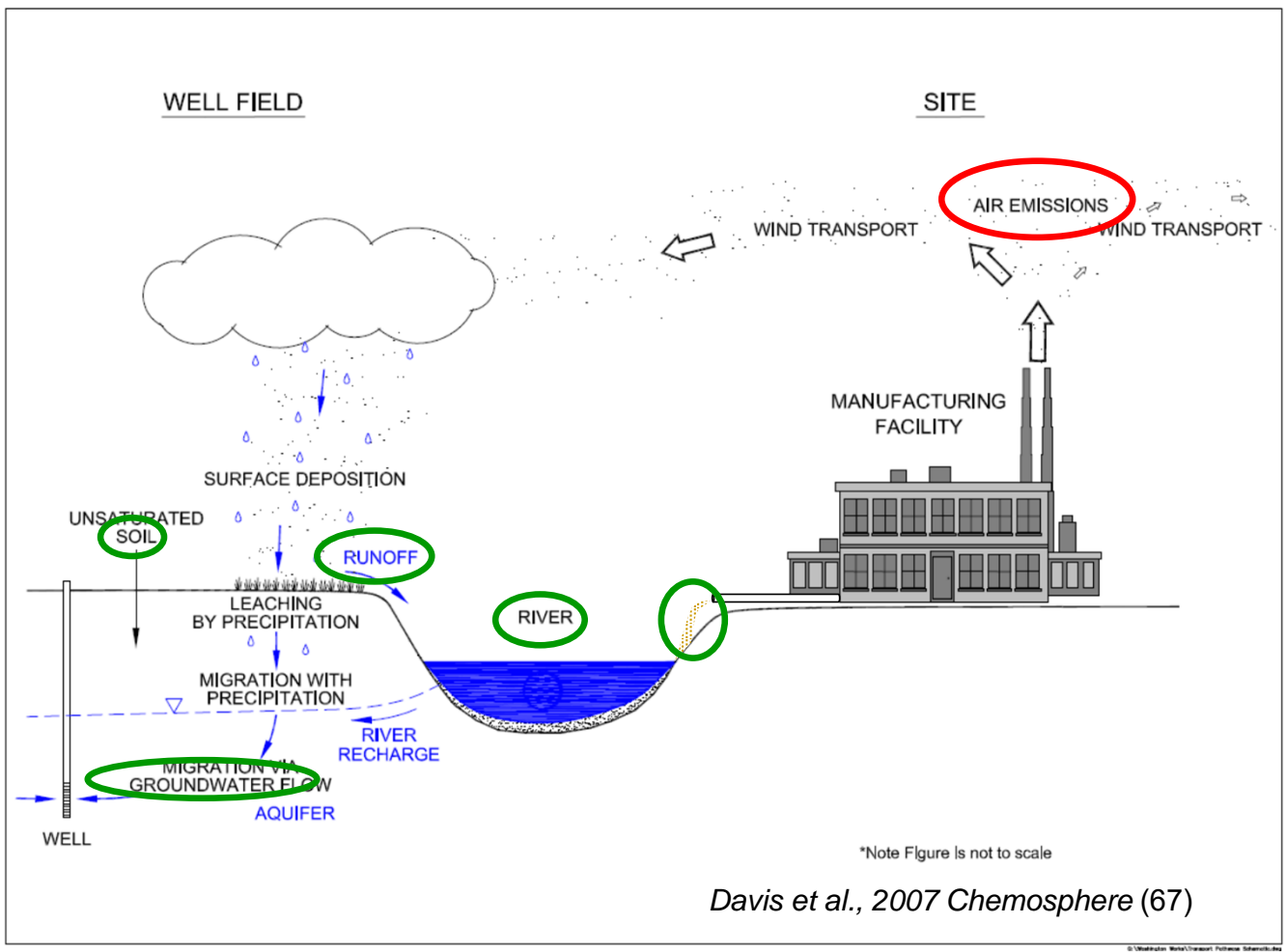
Conceptual Model of APFO Emission



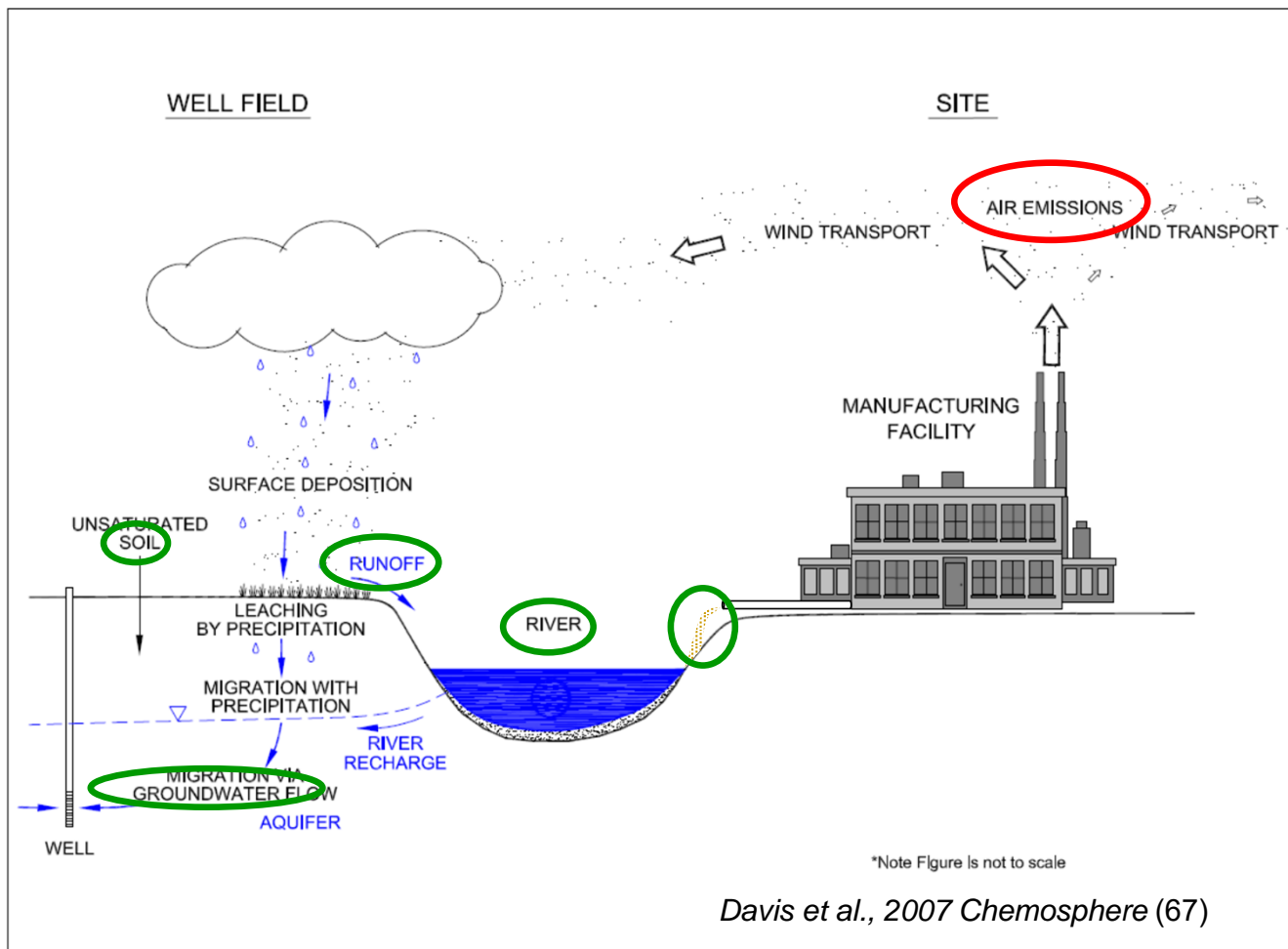
Conceptual Model of APFO Emission



Conceptual Model of APFO Emission



Conceptual Model of APFO Emission



HRMS Analysis to compare Samples:

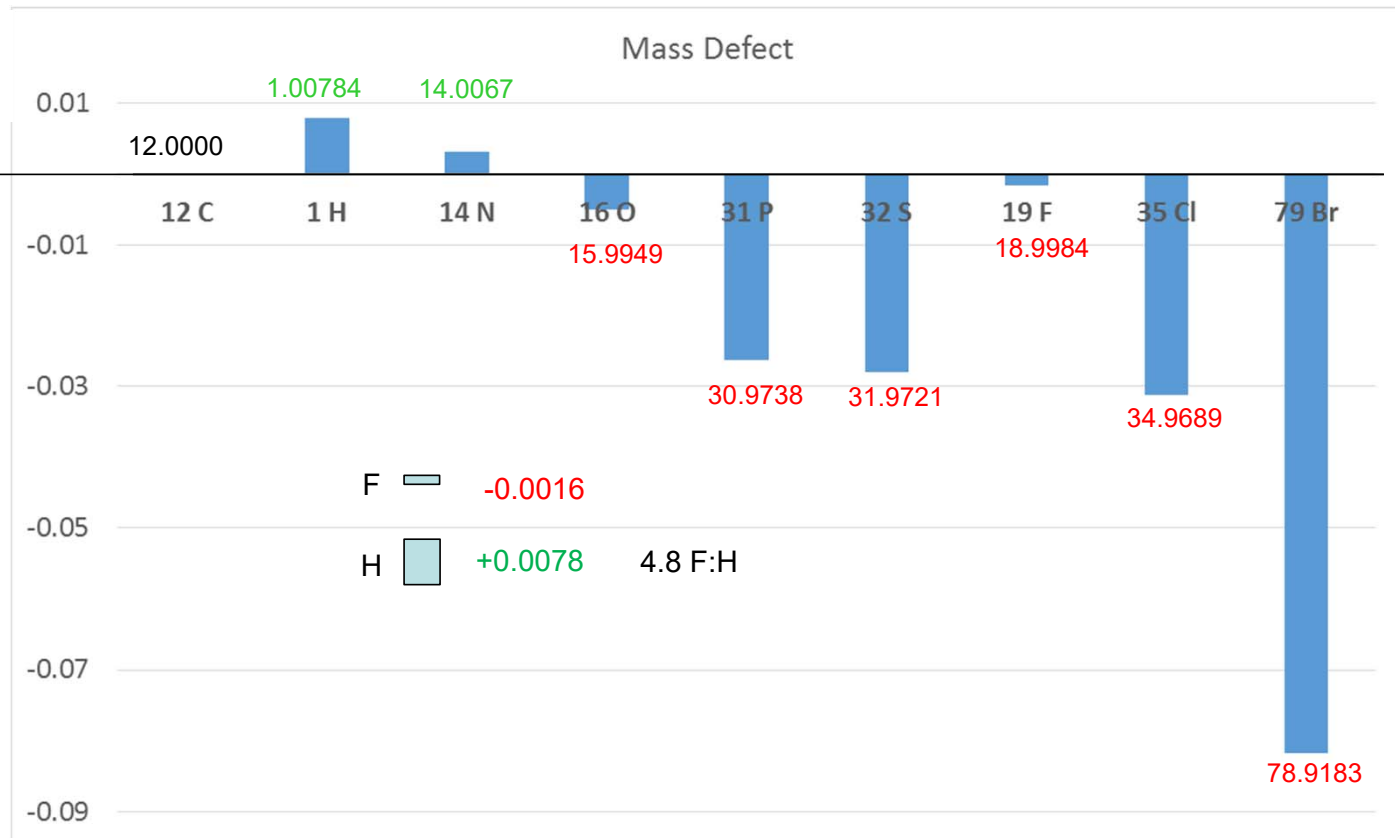
Upstream vs. downstream
Pretreatment vs post-treatment
Close vs distant
Upwind vs downwind
Surface vs deep
Etc....

Isotope Signatures: Negative Mass Defect

SOME KEY TOOLS

Positive Mass Defect

Negative Mass Defect



Isotope Signatures: Negative Mass Defect

SOME KEY TOOLS

Positive Mass Defect

Negative Mass Defect

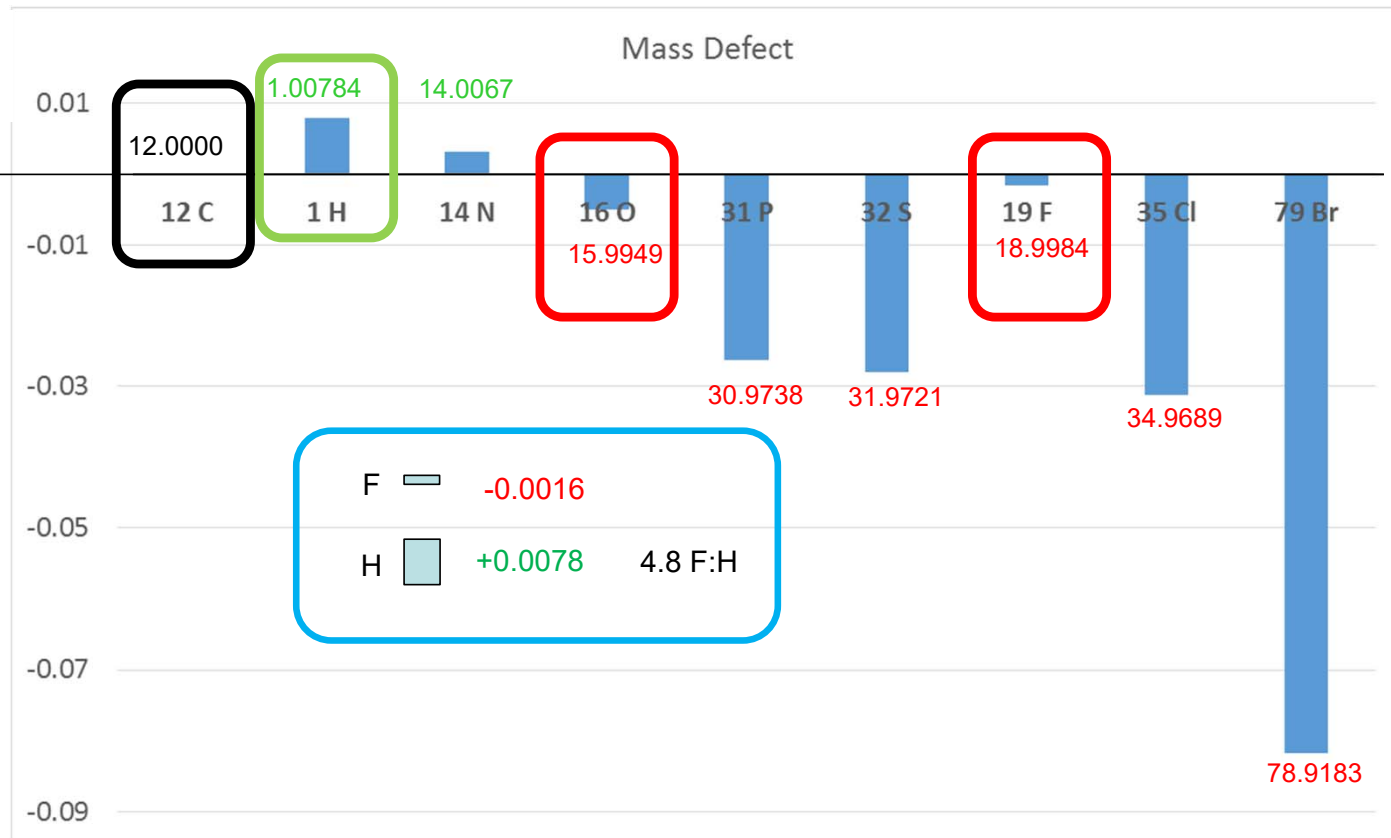


Isotope Signatures: Negative Mass Defect

SOME KEY TOOLS

Positive Mass Defect

Negative Mass Defect

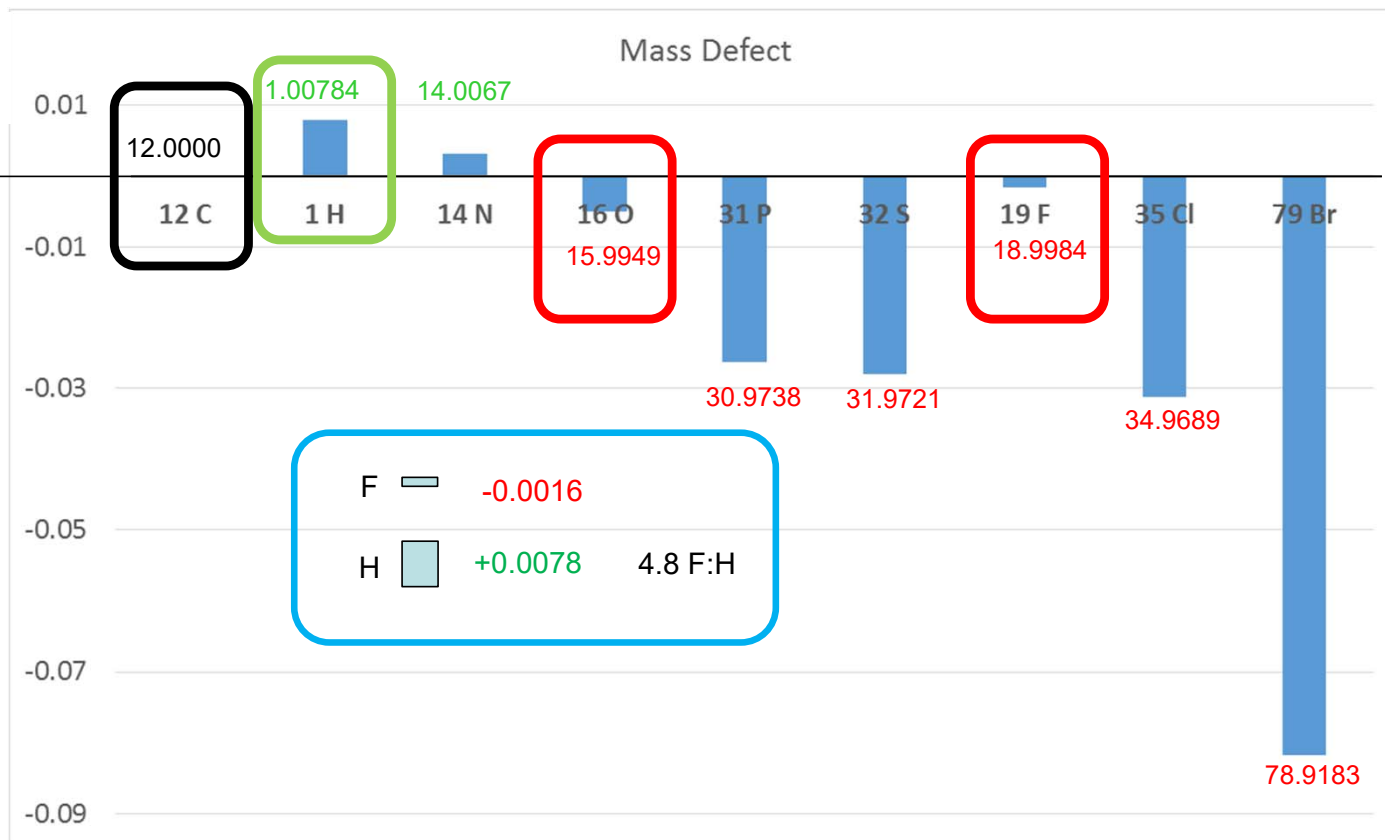


Isotope Signatures: Negative Mass Defect

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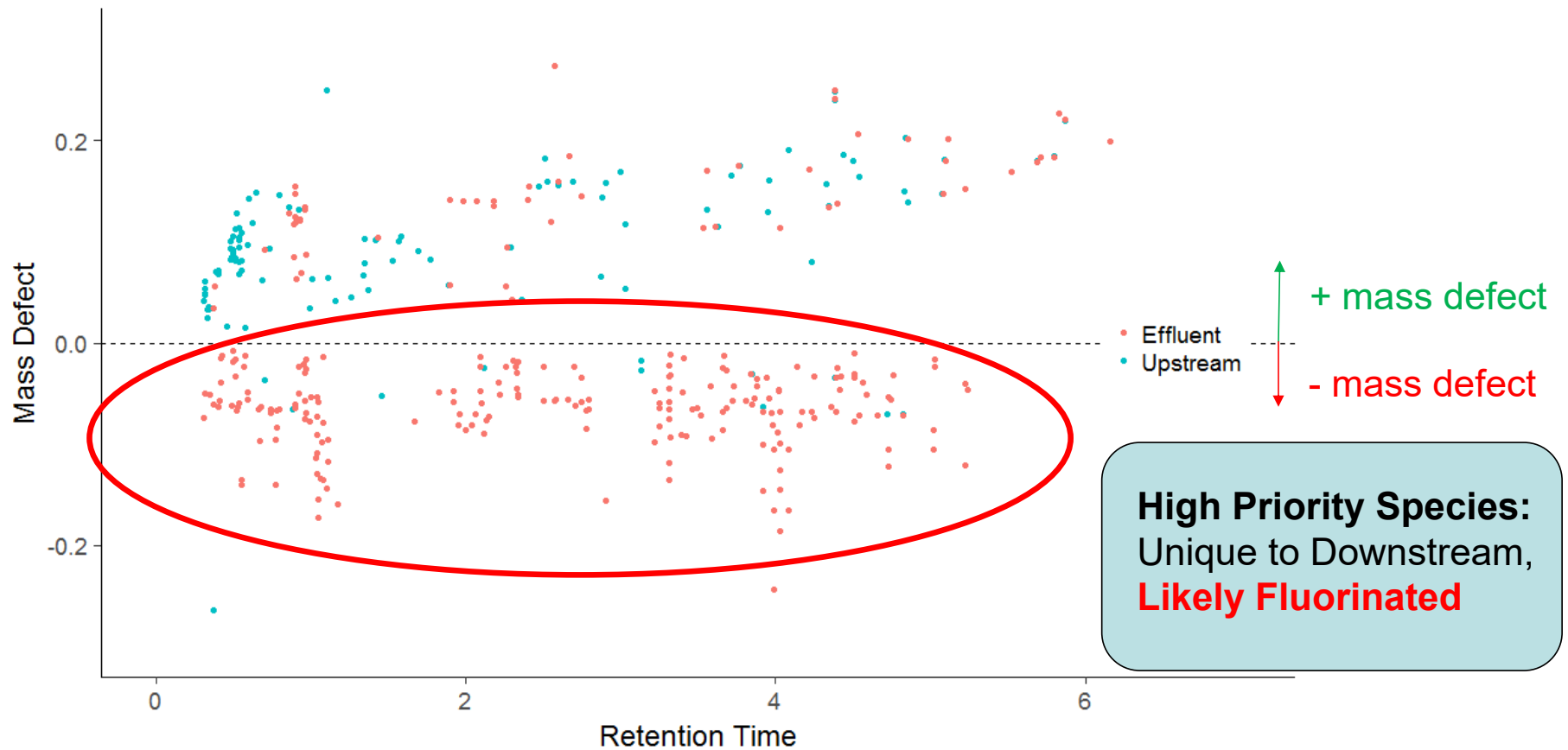


Octanoic Acid
(C₈H₁₆O₂)
144.1150

PFOA
(C₈HF₁₅O₂)
413.9773

Cape Fear River Sample Comparison

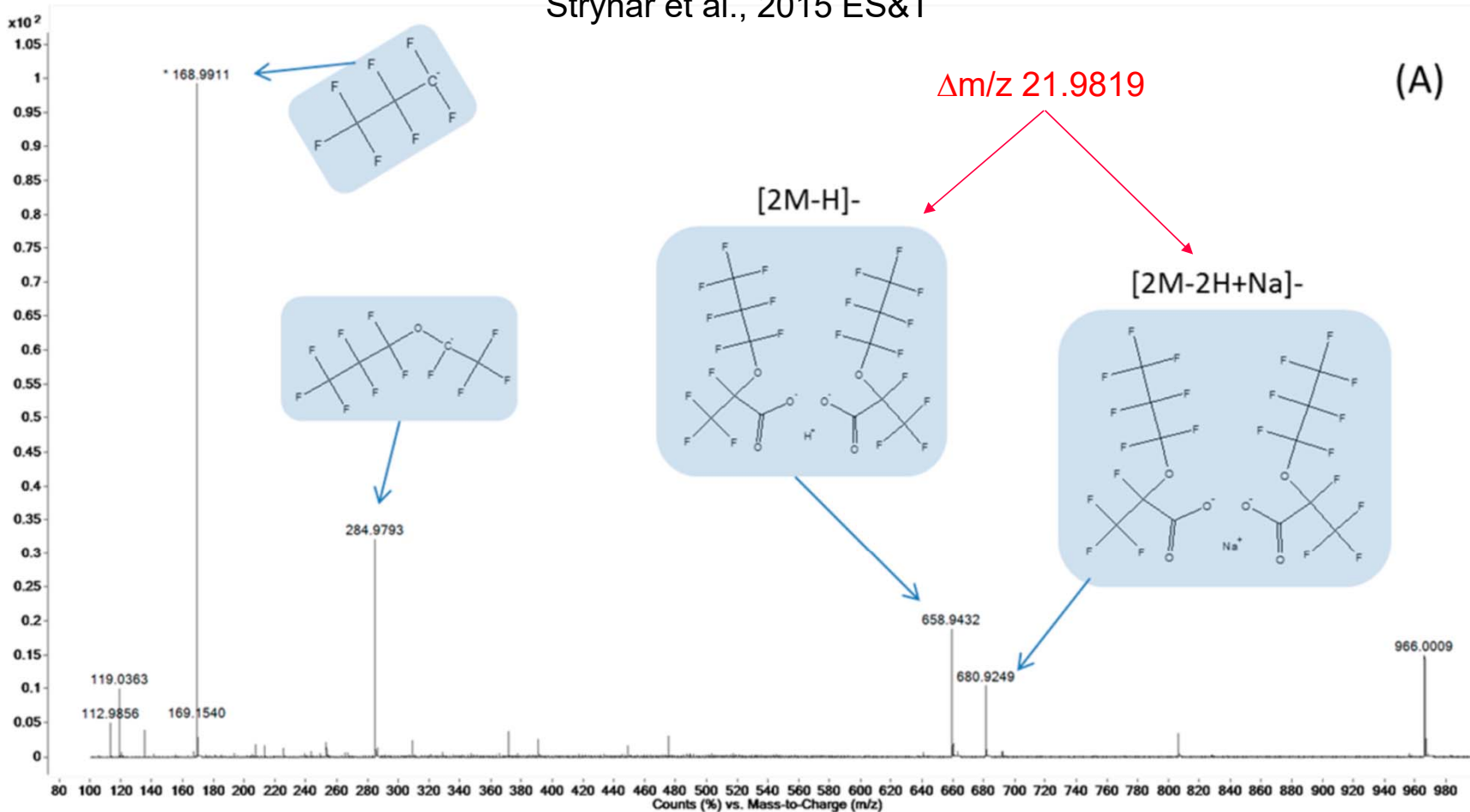
SOME KEY TOOLS



Gas Phase *n*-mers (HFPO-DA)

SOME KEY TOOLS

Strynar et al., 2015 ES&T

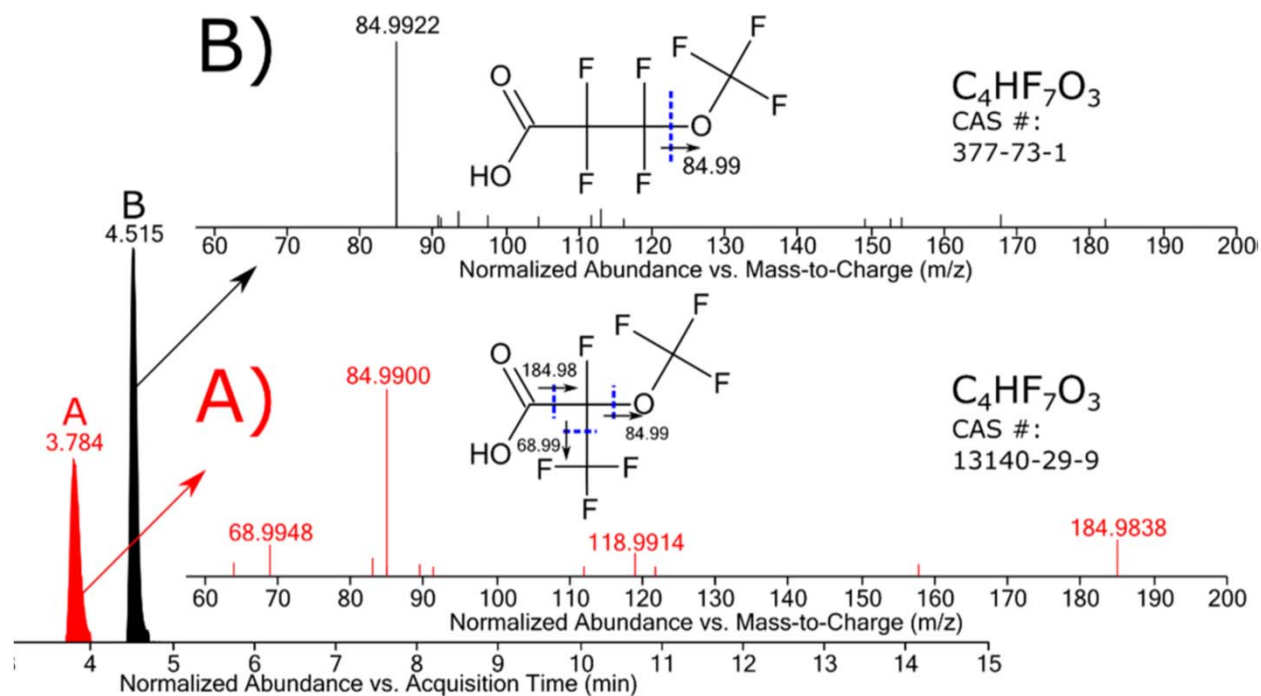


(A)

Have also seen:
 K^+
 NH_4^+
Trimers
tetramers

Diagnostic Fragment Ions contd.

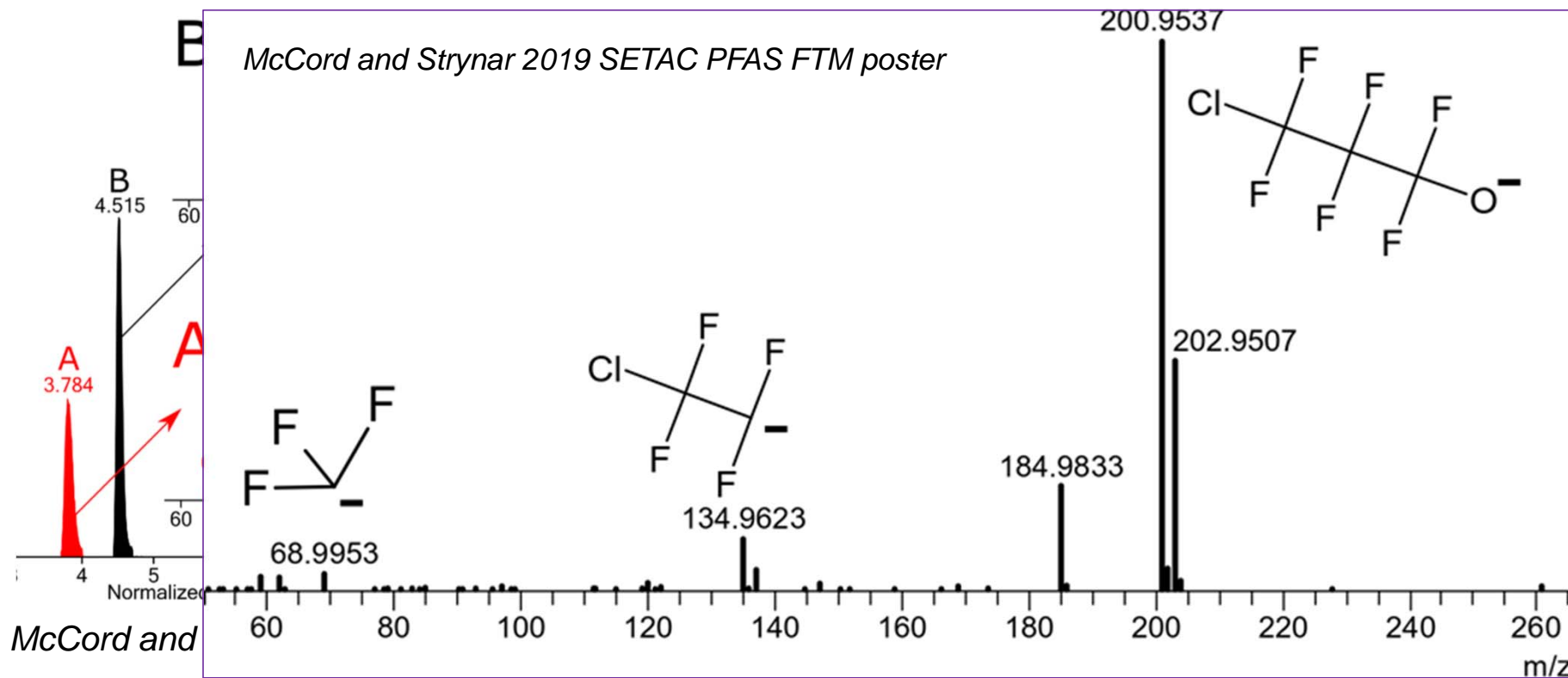
SOME KEY TOOLS



McCord and Strynar ES&T 2019

Diagnostic Fragment Ions contd.

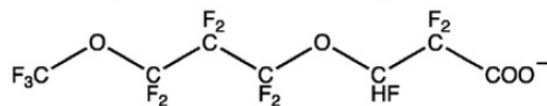
SOME KEY TOOLS



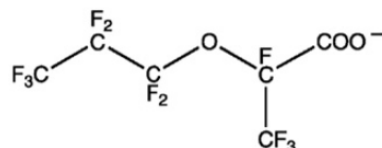
Post PFOA Stewardship Agreement

Fluoropolymer manufacture

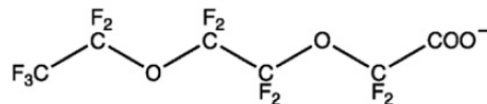
ADONA (CAS No. 958445-44-8)



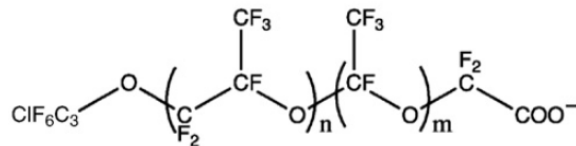
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Asahi's product (CAS No. 908020-52-0)

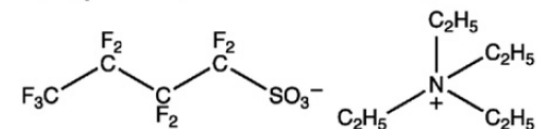


Solvay's product (CAS No. 329238-24-6)

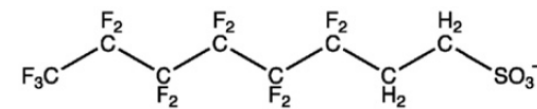


Metal plating

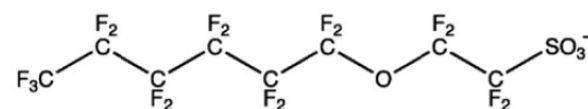
N(Et)₄-PFBS (CAS No. 25628-08-4)



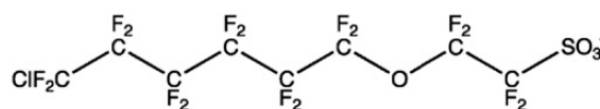
6:2 FTSA (CAS No. 27619-97-2)



F-53 (CAS No. 754925-54-7)



F-53B (CAS No. 73606-19-6)

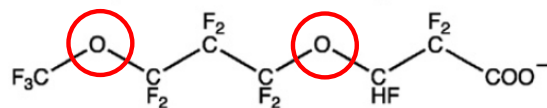


Post PFOA Stewardship Agreement

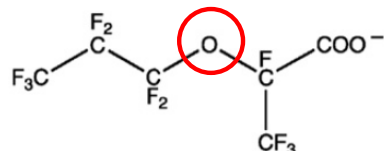
Ethers/Polyethers

Fluoropolymer manufacture

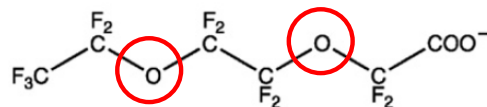
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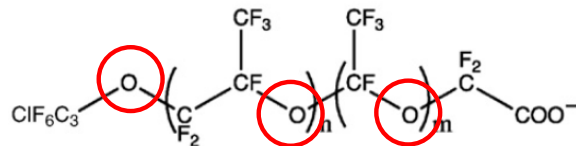
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Asahi's product (CAS No. 908020-52-0)

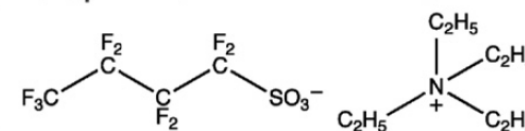


Solvay's product (CAS No. 329238-24-6)

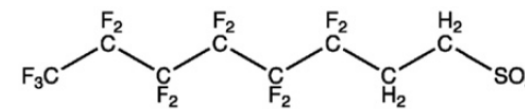


Metal plating

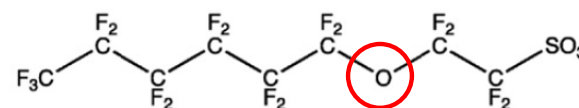
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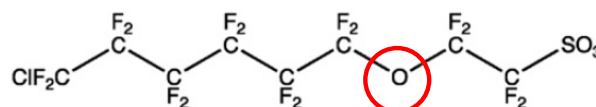
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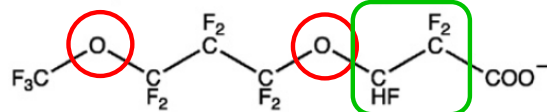
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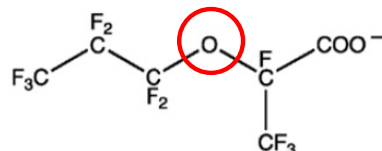
Polyfluorinated

Fluoropolymer manufacture

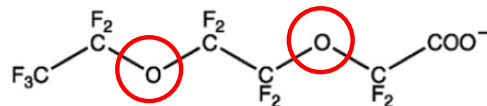
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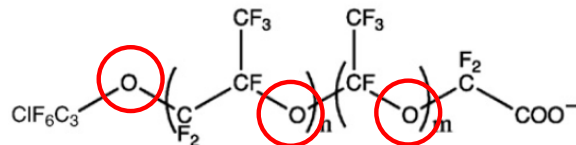
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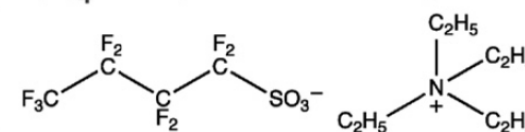


Solvay's product (CAS No. 329238-24-6)

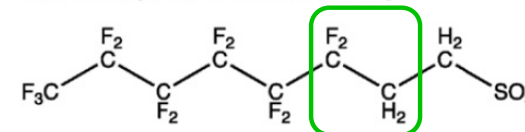


Metal plating

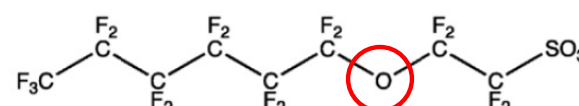
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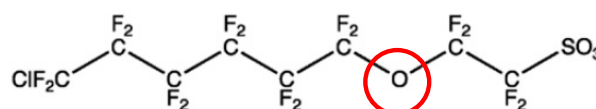
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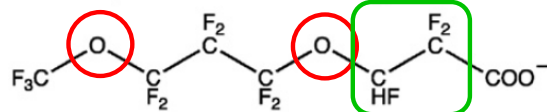
Ethers/Polyethers

Polyfluorinated

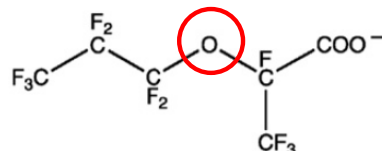
Chlorinated

Fluoropolymer manufacture

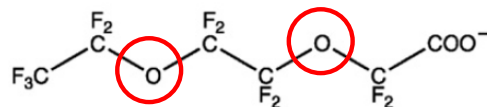
ADONA (CAS No. 958445-44-8)



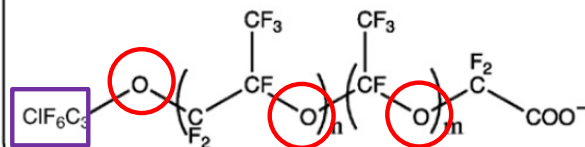
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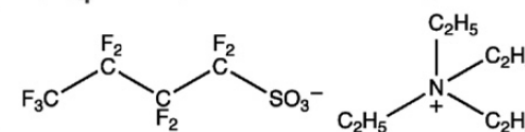


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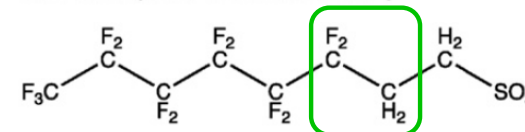


Metal plating

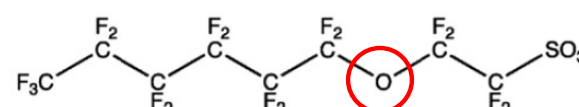
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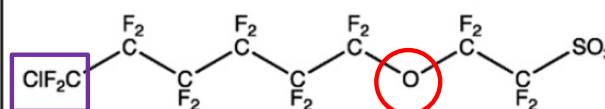
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F-53B (CAS No. 73606-19-6)



Identification of Novel Perfluoroalkyl Ether Carboxylic Acids (PFECAs) and Sulfonic Acids (PFESAs) in Natural Waters Using Accurate Mass Time-of-Flight Mass Spectrometry (TOFMS)

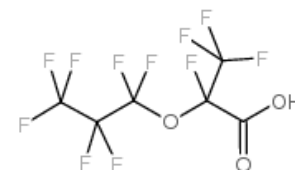
Mark Strynar,^{*,†} Sonia Dagnino,^{‡,‡} Rebecca McMahan,^{‡,‡} Shuang Liang,^{‡,‡} Andrew Lindstrom,[†] Erik Andersen,[†] Larry McMillan,[§] Michael Thurman,^{||} Imma Ferrer,^{||} and Carol Ball[⊥]

Table 1. Accurate Mass of Polyfluorinated Compounds and In-Source Artifacts Found in Extracted Water Samples

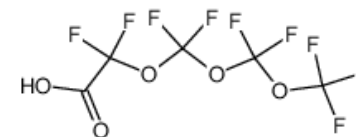
number	formula	CAS no.	name	[M] ^a	[M - H] ⁻ m/z	[2M - 2H + Na] ⁻ m/z	[2M - H] ⁻ m/z
Monoether PFECAs							
1	C ₃ HF ₅ O ₃			179.9846	178.9773	380.9438	358.9619
2	C ₄ HF ₇ O ₃			229.9813	228.9740	480.9372	458.9553
3	C ₅ HF ₉ O ₃	863090-89-5		279.9782	278.9709	580.9310	558.9491
4	C ₆ HF ₁₁ O ₃	13252-13-6	undecafluoro-2-methyl-3-oxahexanoic acid	329.9750	328.9677	680.9247	658.9427
5	C ₇ HF ₁₃ O ₃			379.9718	378.9645	780.9182	758.9363
6	C ₈ HF ₁₅ O ₃			429.9686	428.9613	880.9118	858.9299
Polyether PFECAs							
7	C ₇ HF ₁₃ O ₇	39492-91-6	perfluoro-3,5,7,9,11-pentaoxadodecanoic acid	443.9515	442.9442	908.8776	886.8957
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PFESAs							
11	C ₇ HF ₁₃ O ₅ S	66796-30-3 ^b		443.9337	442.9264		
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3 Found in human serum Wilmington, NC (Kotlarz *EHP* accepted 2020)

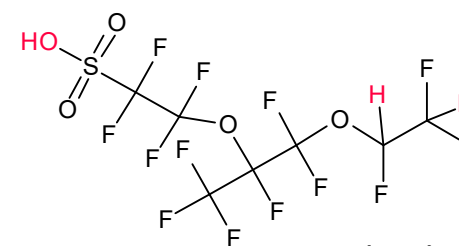
Example Structures



Monoether (6):
GenX



Polyethers (4):



Polyethers
sulfonates (2):

Identification of Novel Perfluoroalkyl Ether Carboxylic Acids (PFECAs) and Sulfonic Acids (PFESAs) in Natural Waters Using Accurate Mass Time-of-Flight Mass Spectrometry (TOFMS)

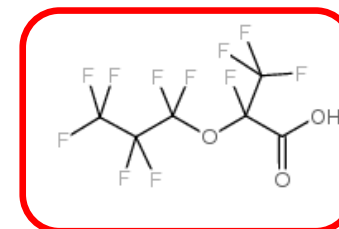
Mark Strynar,^{*,†} Sonia Dagnino,^{‡,‡} Rebecca McMahan,^{‡,‡} Shuang Liang,^{‡,‡} Andrew Lindstrom,[†] Erik Andersen,[†] Larry McMillan,[§] Michael Thurman,^{||} Imma Ferrer,^{||} and Carol Ball[⊥]

Table 1. Accurate Mass of Polyfluorinated Compounds and In-Source Artifacts Found in Extracted Water Samples

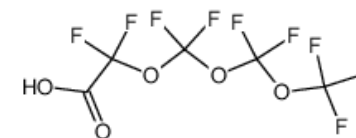
number	formula	CAS no.	name	[M] ^a	[M - H] ⁻ m/z	[2M - 2H + Na] ⁻ m/z	[2M - H] ⁻ m/z
Monoether PFECAs							
1	C ₃ HF ₅ O ₃			179.9846	178.9773	380.9438	358.9619
2	C ₄ HF ₇ O ₃			229.9813	228.9740	480.9372	458.9553
3	C ₅ HF ₉ O ₃	863090-89-5		279.9782	278.9709	580.9310	558.9491
4	C ₆ HF ₁₁ O ₃	13252-13-6	undecafluoro-2-methyl-3-oxahexanoic acid	329.9750	328.9677	680.9247	658.9427
5	C ₇ HF ₁₃ O ₃			379.9718	378.9645	780.9182	758.9363
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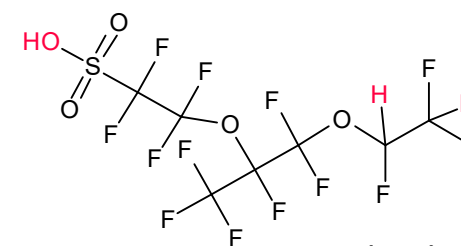
Example Structures



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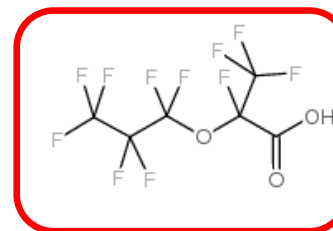
Mark Strynar,^{*,†} Sonia Dagnino,^{†,‡} Rebecca McMahan,^{†,‡} Shuang Liang,^{†,‡} Andrew Lindstrom,[†] Erik Andersen,[†] Larry McMillan,[§] Michael Thurman,^{||} Imma Ferrer,^{||} and Carol Ball[⊥]

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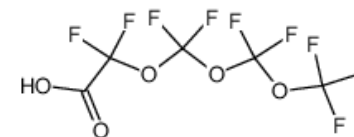
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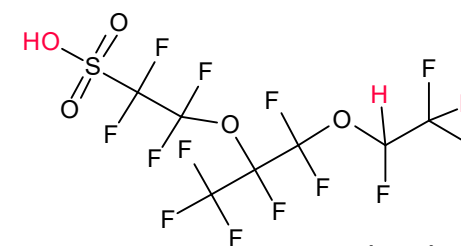
Example Structures



Monoether (6):
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Polyethers (4):



Polyethers
sulfonates (2):

Data Mining



Trends

Clustering

Fingerprinting

Previously Undiscovered

McCord and Strynar ES&T
DOI: 10.1021/acs.est.8b06017
Environ. Sci. Technol. 2019, 53, 4717–4727

Data Mining

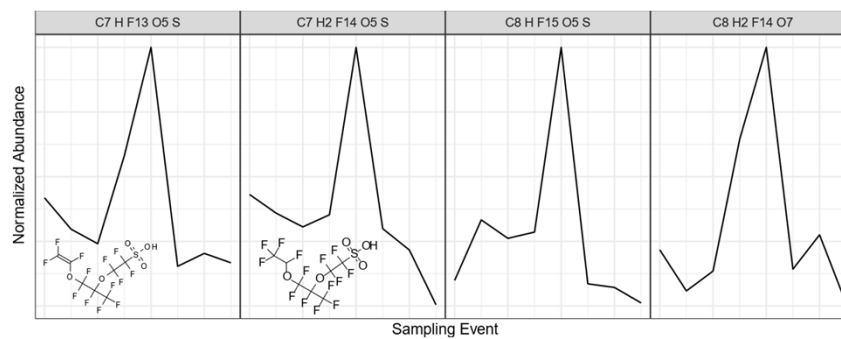


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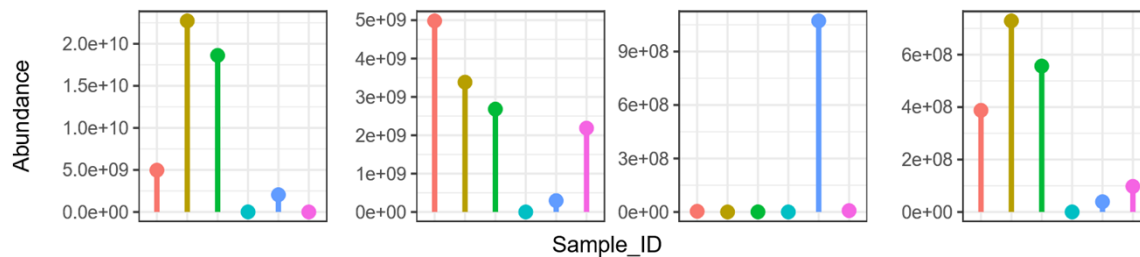


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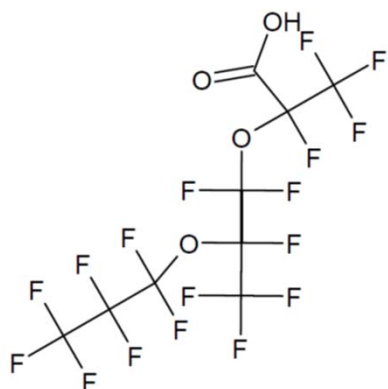


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DOI: 10.1021/acs.est.8b06017
Environ. Sci. Technol. 2019, 53, 4717–4727

Retrospective Analysis

N=20 NC monitoring Wells (M-H)

HFPO-TA CAS 13252-14-7
DTXSID00892442



Molecular Formula: $C_9HF_{17}O_4$
Monoisotopic Mass: 495.960338 Da
[M-H]⁻: 494.953061 Da

Pan et al., 2017 ES&T

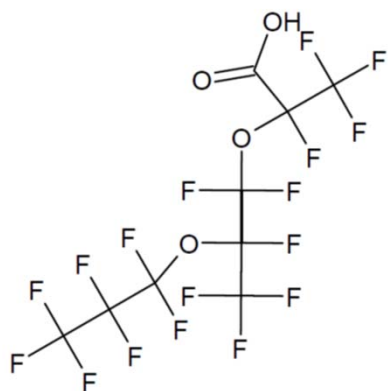
DOI: [10.1021/acs.est.7b02259](https://doi.org/10.1021/acs.est.7b02259)

Xiaoqing River China near fluoropolymer production facility
Found in water, carp and nearby human serum

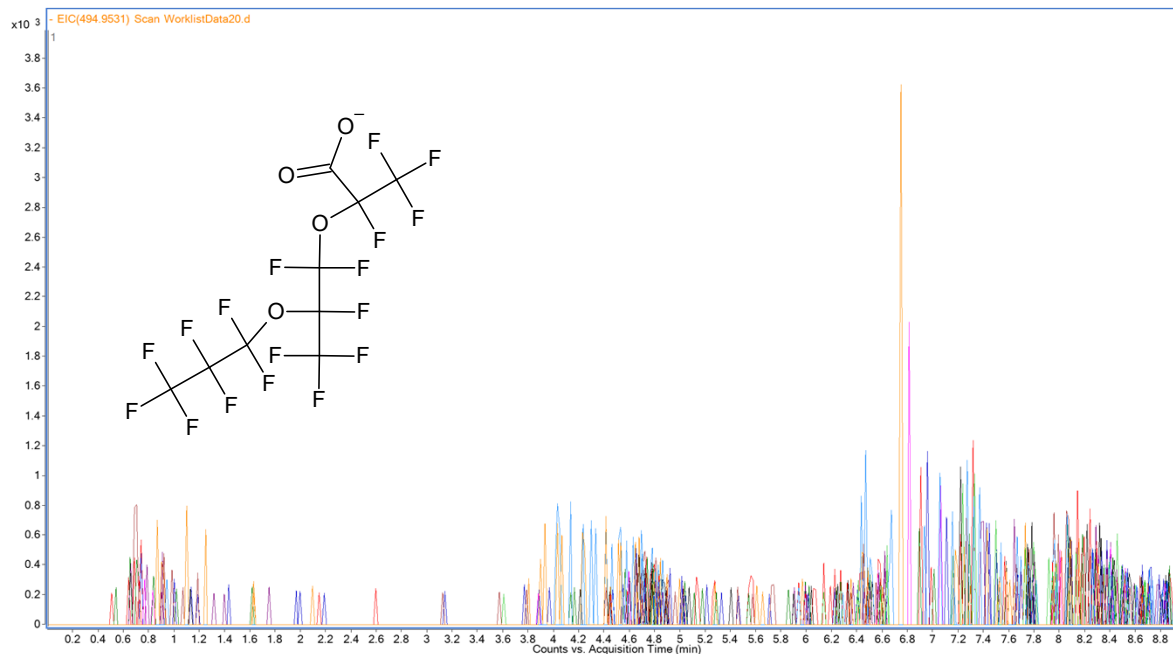
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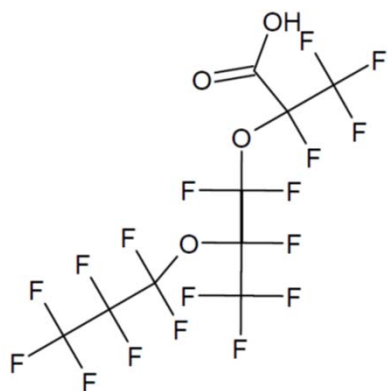


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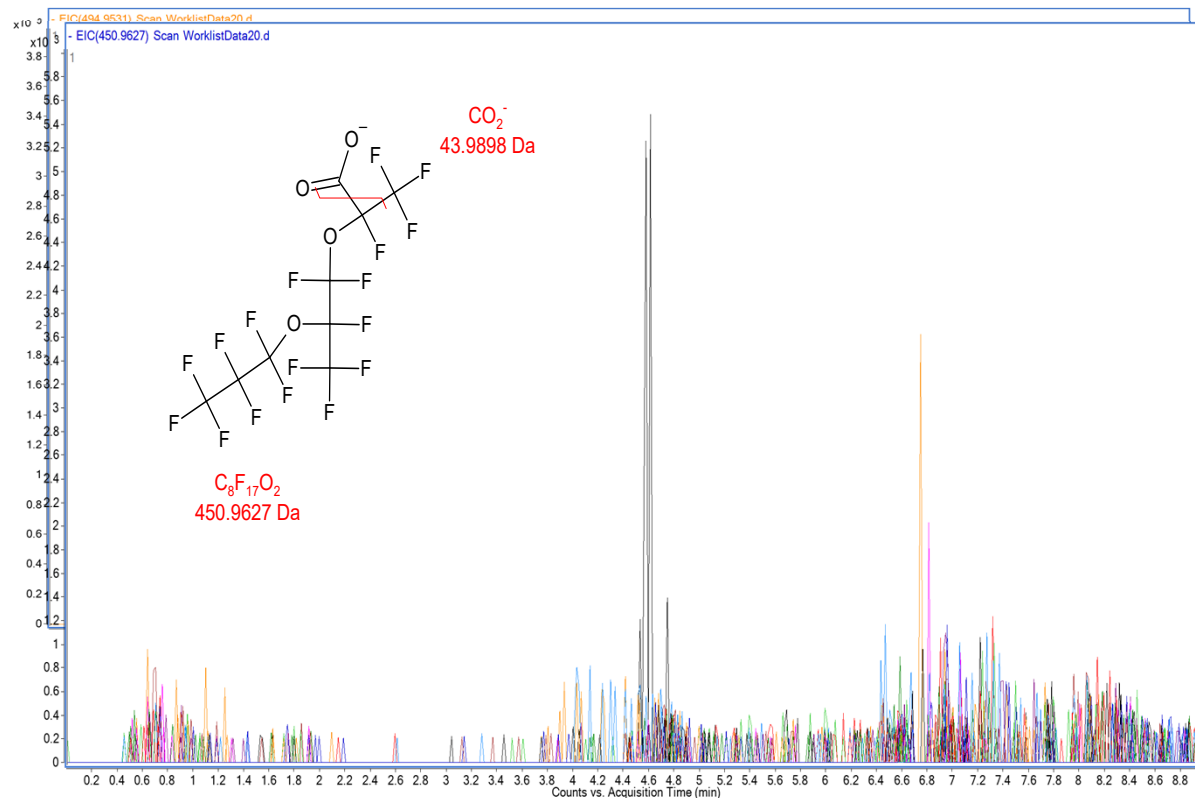
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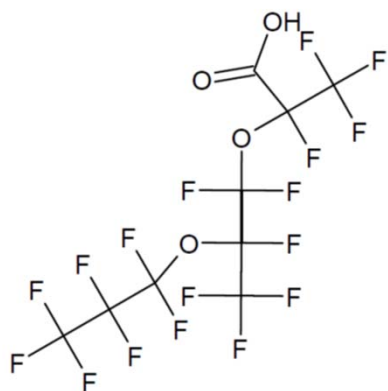
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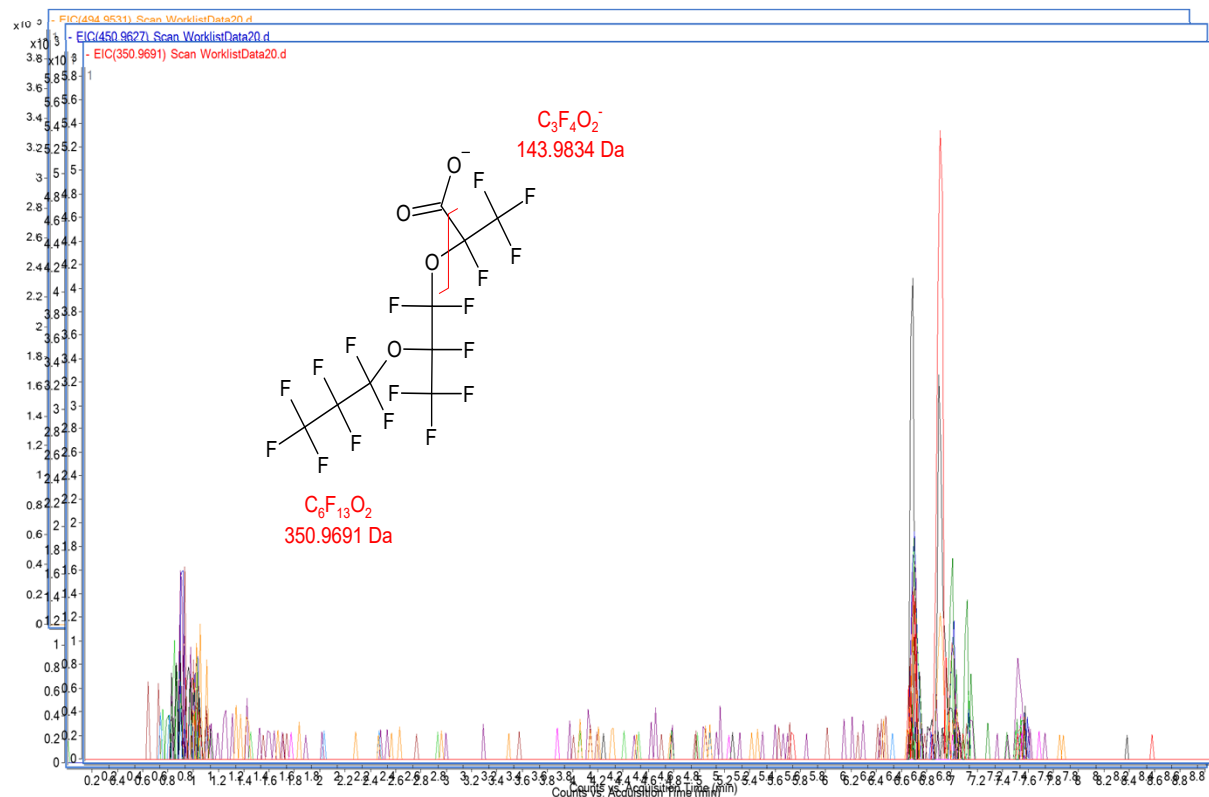


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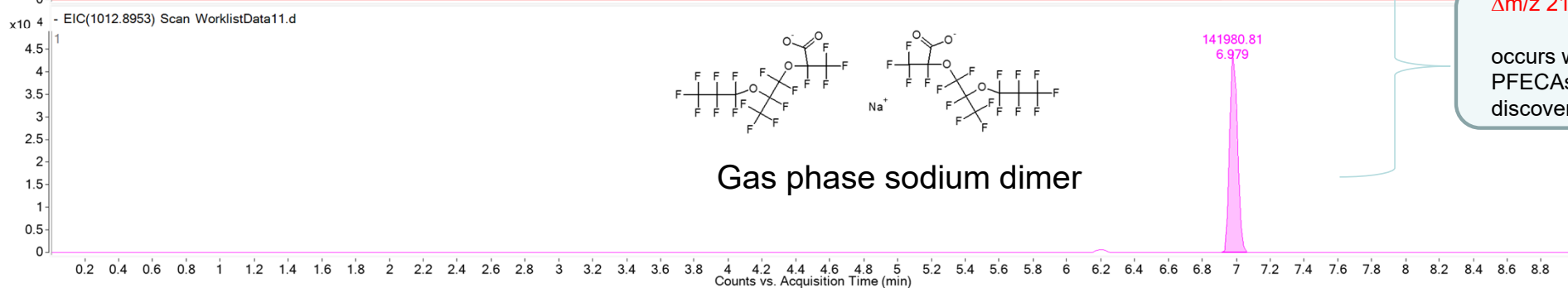
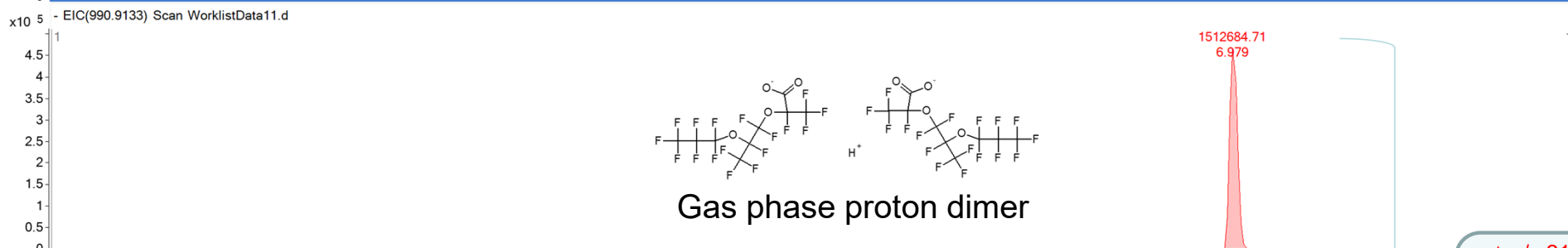
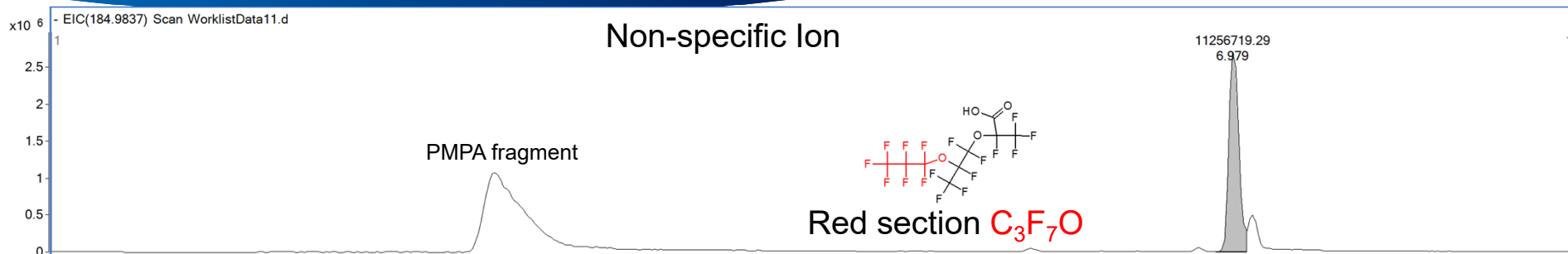
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Retrospective Analysis

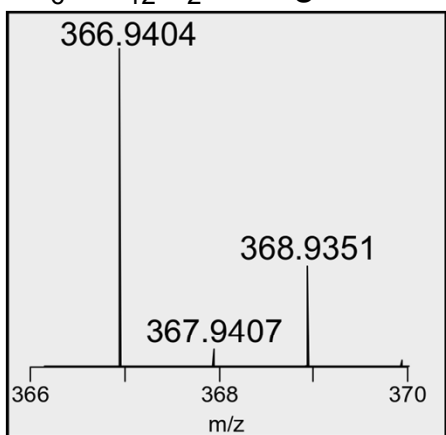
Three TOFMS ions
consistent with HFPO-TA



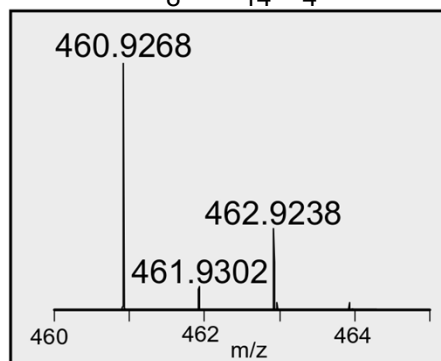
Novel Compound Identification

$$460.9268 * 2 + 1.00784 = 922.8614$$

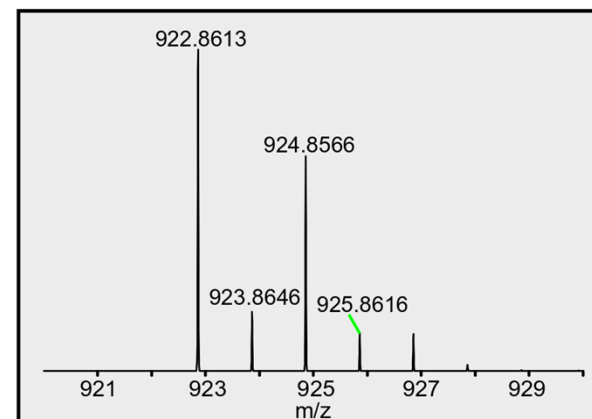
$C_6ClF_{12}O_2^-$ Fragment?



$C_8ClF_{14}O_4$

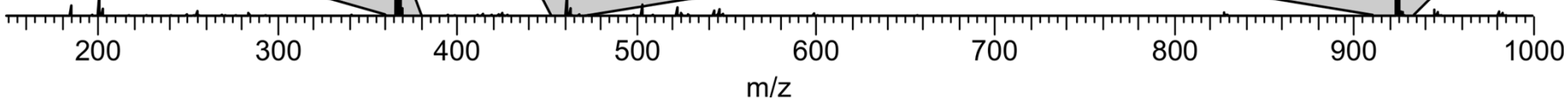


Dimer

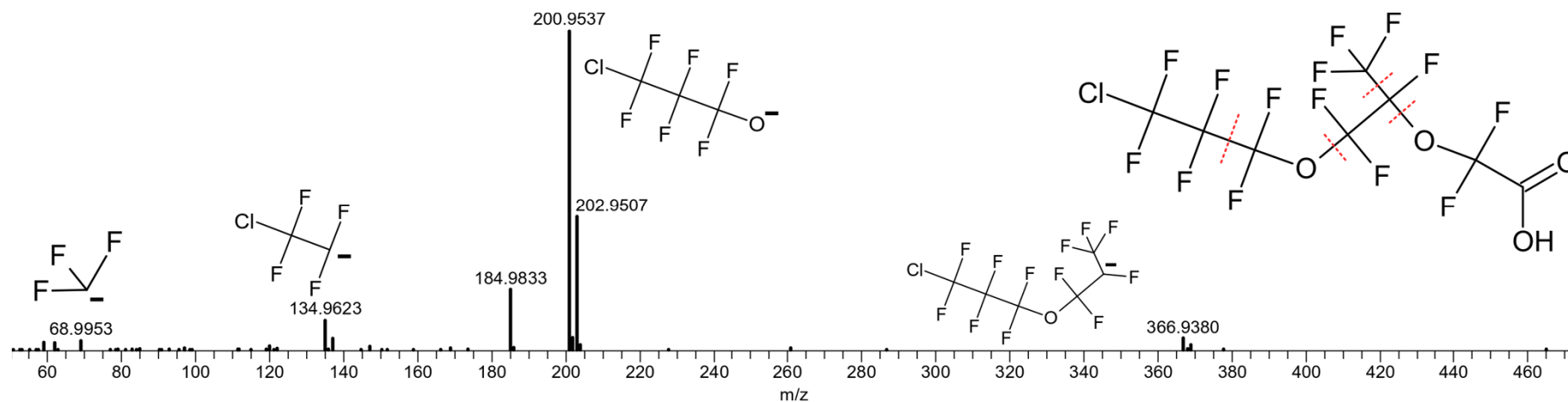


C_3ClF_6O

200.9548



Structural Elucidation by MS/MS

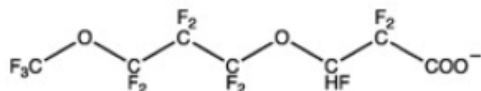


- Unequivocal assignment of terminal Cl and ether positions based on MS/MS experiments
- Confirmation of Dimers and in-source fragments from prior slide, with additional experiments (not show)

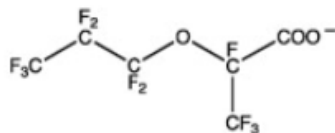
Literature Support

Fluoropolymer manufacture

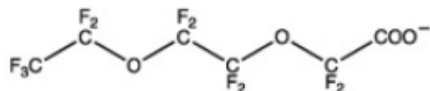
ADONA (CAS No. 958445-44-8)



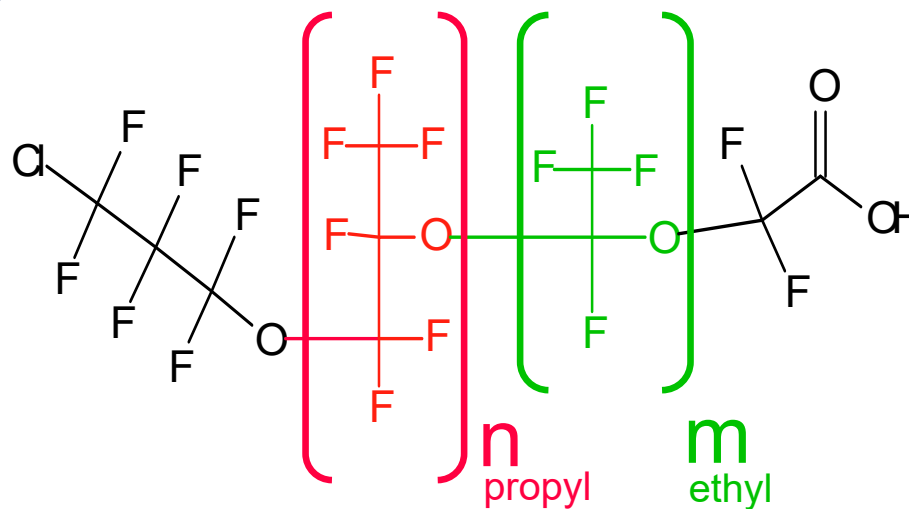
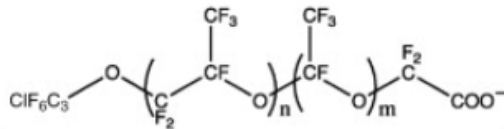
GenX (CAS No. 62037-80-3)



Asahi's product (CAS No. 908020-52-0)



Solvay's product (CAS No. 329238-24-6)

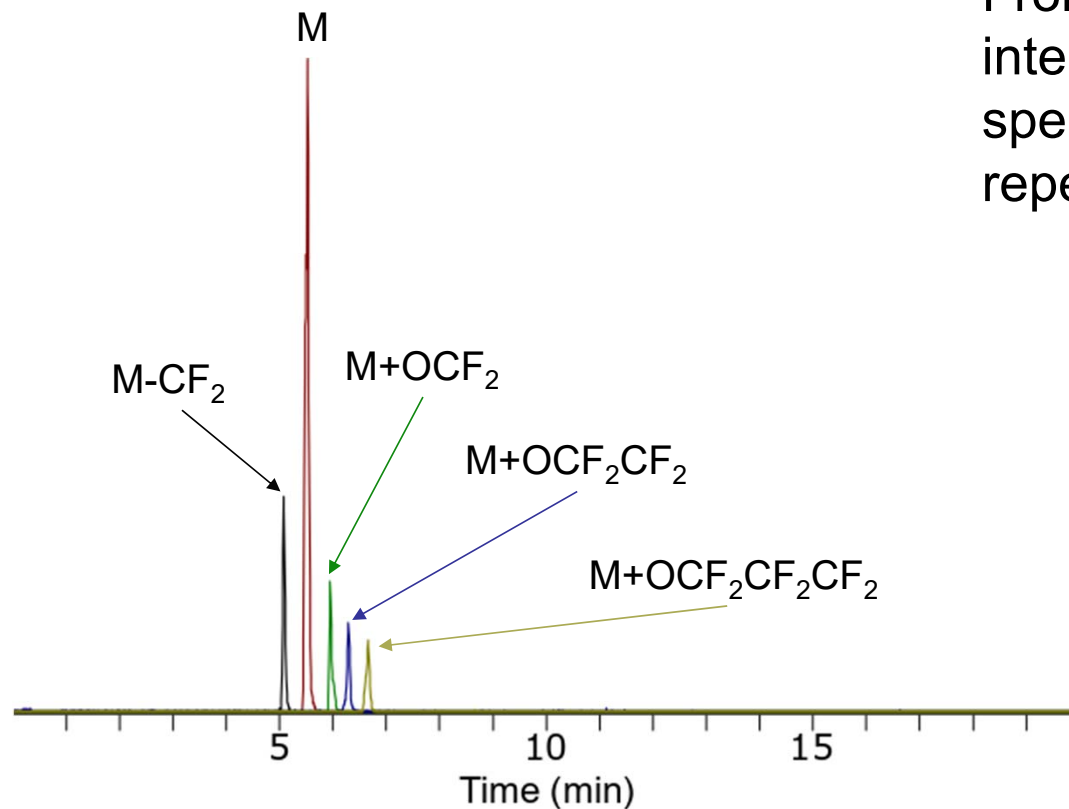


Chloro perfluoro polyether carboxylic acids CIPFPECAs(n,m)

Wang, Z., et al. (2013). *Environ. Int.* **60**: 242.

EFSA J, 8 (2) (2010), p. 1519, [10.2903/j.efsa.2010.1519](https://doi.org/10.2903/j.efsa.2010.1519)

Homologous Fluoropolymer Series

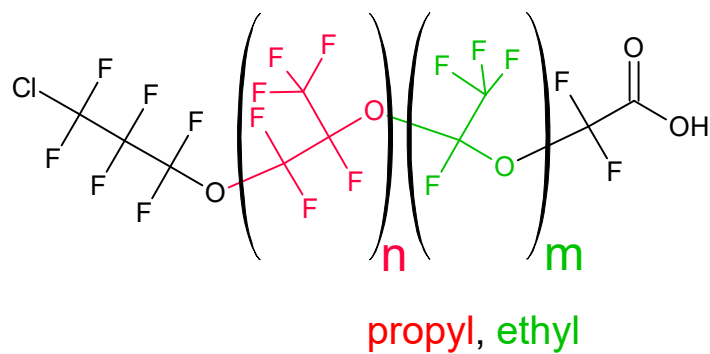


From single identified mass of interest we identified additional species related by CF₂ repeating units

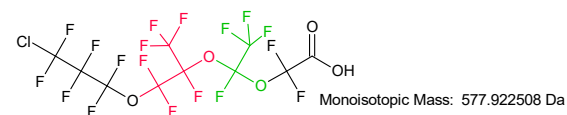
Washington et al., 2020 Nontargeted mass-spectral detection of chloroperfluoropolyether carboxylates in New Jersey soils *Science*
DOI: [10.1126/science.aba7127](https://doi.org/10.1126/science.aba7127)

Structural Homologs

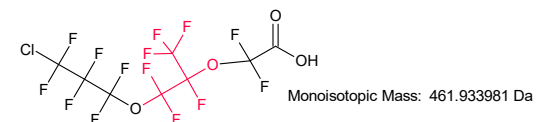
CAS 329238-24-6



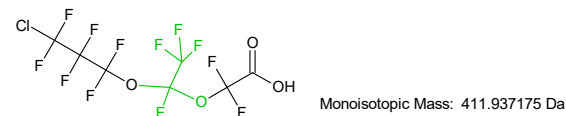
CIPFECA 1,1



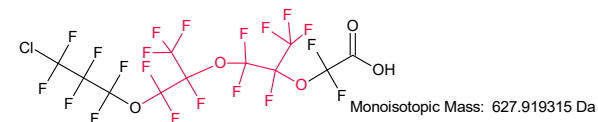
CIPFECA 1,0



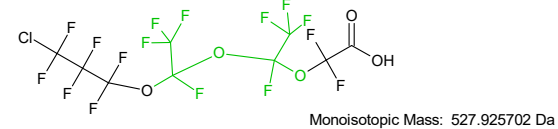
CIPFECA 0,1



CIPFECA 2,0



CIPFECA 0,2



Some Take Home Points



- You only see what you are looking for
- HRMS has a role to discover the undiscovered
- HRMS should be used in concert with LRMS or other methods
- Retrospective analysis is possible



Questions?

Contact Information
strynar.mark@epa.gov

The views expressed in this presentation are those of the author(s) and do not necessarily represent the views or policies of the U.S. Environmental Protection Agency.

U.S. Environmental Protection Agency